

## **Cadmium Toxicity to Photosynthesis and Associated Electron Transport System of *Nostoc linckia***

Yasmin Husaini,<sup>1</sup> Ashok Kumar Singh,<sup>2</sup> and L. C. Rai<sup>1</sup>

<sup>1</sup>Centre of Advanced Study in Botany and <sup>2</sup>School of Biotechnology, Banaras Hindu University, Varanasi-221005, India

The significance of cadmium as an aquatic contaminant has long been realized. It is during the last few years that much attention has been focussed on growth, O<sub>2</sub> evolution, carbon fixation, nitrogenase activity, ultrastructural changes and localization of Cd in cell cytoplasm (see Rai *et al.* 1981; Vymazal 1987; Rai *et al.* 1989). Notwithstanding, the impact of Cd on structural and functional characteristics of phytoplanktons in natural systems has also been studied (Wong 1987).

It is worth mentioning that despite much emphasis already given to understand the toxicity of Cd nothing is known about the site of its action in cyanobacteria and algae. Therefore the present work has been undertaken not only to find out the site of action of Cd in cyanobacteria but also to know the mechanism of inhibition of photosynthetic electron transport, a process responsible for the generation of ATP and NADPH, which are essential for carbon fixation. The present study compares the sensitivities of photosystem I (PS I), photosystem II (PS II) and redox coupling between the two photosystems of *Nostoc linckia* exposed to different concentrations of cadmium.

### **MATERIALS AND METHODS**

*Nostoc linckia* was grown axenically in modified Chu 10 medium (Gerloff *et al.* 1950) buffered with 4 mM Tris/HCl buffer (pH 8) under 14.4 Wm<sup>-2</sup> light intensity and a 14:10 h photoperiod at 26±2°C. Stock solution of CdCl<sub>2</sub> was filter sterilized by passing through Millipore membrane filters (0.45 µM) before adding to the culture medium. Three concentrations of test metals viz. (i) effective concentration (EC<sub>50</sub>), (ii) one concentration above and (iii) one concentration below EC<sub>50</sub> were taken to study their effects.

-----  
Corresponding author: L.C. Rai

The protein content of the cells was determined by the method of Lowry et al. (1951). Photosynthetic  $O_2$  evolution of exponentially growing Nostoc cells ( $300 \mu g$  protein  $ml^{-1}$ ) was determined in terms of  $\mu mole O_2 mg$  protein  $h^{-1}$  by a calibrated Clark type  $O_2$  electrode, enclosed in a  $10 ml$  airtight reaction vessel illuminated with  $12.0 W m^{-2}$  light at  $25 \pm 2^\circ C$ . This was connected to an oxygen analyzer (Universal Biochem. Model M76T, India). Cell free thylakoid membrane of Nostoc linckia was prepared following the method of Lien (1976).

Hill reaction assays were carried out within three hours of membrane preparation by polarographic method (Lien 1976) and expressed in terms of either evolution or consumption of  $O_2$ .

Reaction 1.  $H_2O \rightarrow p\text{-benzoquinone (PBQ)}$  (assayed as  $O_2$  evolution).

Reaction 2.  $H_2O \rightarrow \text{Ferricyanide}$  (assayed as  $O_2$  evolution).

Reaction 3. Ascorbate - 2,6-dichlorophenol-indophenol sodium salt (DCPIP)  $\rightarrow$  methyl viologen (MV) (assayed as  $O_2$  consumption).

The addition of 3-(3,4 dichlorophenyl)-1,1-dimethylurea (DCMU) ( $0.01 ml, 5mM$ ) blocked the PS II electron transport.

Carbon fixation was determined by measuring the uptake of  $^{14}C$  from  $NaH^{14}CO_3$  by the method of Rai and Raizada (1986) and expressed as dpm (disintegration per minute).

The total ATP content was measured by luciferin-luciferase assay using LKB-1250 Luminometer. The ATP was extracted by using 4% TCA supplemented with 2 mM EDTA.

## RESULTS AND DISCUSSION

The effect of different concentrations of cadmium on  $O_2$  evolution,  $^{14}C$  incorporation and ATP content of Nostoc linckia is given in Table 1. Cadmium inhibited the electron transport chain,  $^{14}C$  uptake and ATP content in a concentration dependent manner. The test cyanobacterium pre-exposed to different doses of cadmium ( $0.01$  to  $0.05 \mu g ml^{-1}$ ) and incubated for 1 h under normal growth conditions, showed higher inhibition of carbon assimilation as compared to  $O_2$  evolution. These results are in essential agreement with those of Rai (1989), where  $^{14}C$  incorporation has

been suggested to be the most sensitive parameter in metal toxicity evaluations. The inhibitory action of cadmium on photosynthesis as observed in the present study is due to the inhibition of photosystem I and II (see Table 2).

Reaction 1.  $H_2O \rightarrow PBQ$  is dependent on PS II and does not have an absolute requirement for PS I for an efficient coupling between the two photosystems (assayed in terms of  $O_2$  evolution). This reaction was found sensitive to both low and high concentrations of Cd. A four fold decrease in PS II activity was observed with increase in Cd concentration from 0.01 to 0.05  $\mu g\ ml^{-1}$ . PS I activity (Reaction 3) which was assayed in terms of  $O_2$  consumption showed less inhibition at a low concentration (0.01  $\mu g\ ml^{-1}$ ). However, the toxicity increased by approximately eight fold when Cd concentration was raised from 0.01 to 0.05  $\mu g\ ml^{-1}$ .

Table 1. Effect of different concentrations of cadmium on photosynthetic  $O_2$  evolution,  $^{14}C$  incorporation and ATP content of Nostoc linckia

| Cd concentration<br>( $\mu g\ ml^{-1}$ ) | $O_2$ evolution<br>( $\mu M\ O_2\ mg^{-1}\ protein\ h^{-1}$ ) | $^{14}C$ uptake<br>(dpm $mg^{-1}\ protein\ h^{-1}$ ) | ATP content<br>(ng ATP $mg^{-1}\ protein\ h^{-1}$ ) |
|--|---|--|---|
| Control                                  | 48.2 $\pm$ 1.8*   | 656137 $\pm$ 32885                                   | 1941 $\pm$ 91.6                                     |
| 0.01                                     | 44.2 $\pm$ 2.2 <sup>a</sup><br>(8.2%)**                       | 544453 $\pm$ 32666 <sup>a</sup><br>(17.0%)           | 1771 $\pm$ 27.5 <sup>a</sup><br>(8.7%)              |
| 0.03                                     | 39.0 $\pm$ 1.5 <sup>b</sup><br>(19.0%)                        | 417831 $\pm$ 16704 <sup>b</sup><br>(36.3%)           | 1299 $\pm$ 61.6 <sup>b</sup><br>(33.0%)             |
| 0.05                                     | 31.5 $\pm$ 1.7 <sup>b</sup><br>(34.6%)                        | 275475 $\pm$ 18670 <sup>b</sup><br>(41.9%)           | 1014 $\pm$ 47.0 <sup>b</sup><br>(47.7%)             |

\* Mean  $\pm$  S.D.

\*\* Per cent inhibition

Values significantly different from control have been marked: <sup>a</sup>  $P < 0.05$ , <sup>b</sup>  $P < 0.01$  (student's 't' test).

Reaction 2.  $H_2O \rightarrow$  Ferricyanide (assayed as  $O_2$  evolution which is absolutely dependent on a functional PS, II, PS I and redox coupling between the two photosystems, was slightly inhibited (9.6%) at 0.01  $\mu g\ ml^{-1}$  Cd and approximately 50% at high concentration (0.05  $\mu g\ ml^{-1}$ ). Since ATP and NADPH are the primary requirements for  $CO_2$  fixation, one can reasonably infer that inhibition of PS II activity both at low as well

as high concentrations, and PS I activity at high concentration of Cd was responsible for unavailability of ATP and NADPH and hence for the inhibition of photosynthetic carbon fixation and O<sub>2</sub> evolution. It is evident from Table 1 that ATP content of cyanobacterium was strongly and equally inhibited by Cd as <sup>14</sup>C incorporation. Therefore it becomes clear that reduction in ATP synthesis was responsible for inhibition of <sup>14</sup>C incorporation.

Table 2. Effect of cadmium on Hill activity by membrane preparation from Nostoc linckia-1 (μmole O<sub>2</sub> evolved or consumed mg<sup>-1</sup> protein h<sup>-1</sup>)

| Cd concentration<br>(μg ml <sup>-1</sup> ) | Reaction 1<br>H <sub>2</sub> O → PBQ | Reaction 2<br>H <sub>2</sub> O → Ferri-<br>cyanide | Reaction 3<br>Ascorbate-DCPIP<br>→ MV |
|--|--------------------------------------|--|---------------------------------------|
| Control                                    | 29.85±1.5*                           | 38.29±2.3  | 45.34±1.8                             |
| 0.01                                       | 26.02±1.6 <sup>a</sup><br>(12.8%)**  | 34.60±1.9 <sup>a</sup><br>(9.6%)                   | 43.50±1.9<br>(4.0%)                   |
| 0.03                                       | 20.40±1.0 <sup>b</sup><br>(31.6%)    | 23.50±1.3 <sup>b</sup><br>(38.5%)                  | 37.99±1.9 <sup>a</sup><br>(16.2%)     |
| 0.05                                       | 15.46±0.8 <sup>b</sup><br>(48.2%)    | 17.30±1.0 <sup>b</sup><br>(54.8%)                  | 29.56±1.8 <sup>b</sup><br>(34.8%)     |

\* Mean ± S.D.

\*\* Per cent inhibition

Values significantly different from control have been marked: <sup>a</sup>P<0.05, <sup>b</sup>P<0.01 (student's 't' test).

These results clearly indicate that PS II is the primary site of action of Cd for photosynthetic electron transport as was observed by Vierke and Struckmeier (1977) in Spinach chloroplast and by Samson et al. (1988) for Dunaliella tertiolecta for copper. Since the chloroplast of eukaryotic algae and cyanobacteria are phylogenetically related (Lang 1968) we believe that PS II may be the primary site of action of Cd in cyanobacteria and algae.

**Acknowledgments.** We thank the Head, Department of Botany, Banaras Hindu University, for facilities and the University Grants Commission, New Delhi, for financial support in the form of career award of L.C. Rai.

## REFERENCES

- Gerloff GC, Fitzgerald GP, Skoog F (1950) The isolation, purification and culture of blue-green algae. *Am J Bot* 37: 216-218.

- Lang NJ (1968) The fine structure of blue-green algae. *Ann Rev Microbiol* 2: 15-46.
- Lien S (1976) Hill reaction and phosphorylation with chloroplast preparations from Chlamydomonas reinhardtii. In: Hellebust JA, Craigie JS (eds.) Handbook of Phycological Methods: Physiological Methods, Cambridge University Press, p. 305.
- Lowry OH, Rosebrough NJ, Farr AL, Randall RJ (1951) Protein measurement with the Folin phenol reagent. *J Biol Chem* 193:269-275.
- Rai LC (1989) Silver toxicity in a nitrogen-fixing cyanobacterium. Interaction with chromium, nickel and lead. *Biol Metals* 2: 122-128.
- Rai LC, Gaur JP, Kumar HD (1981) Phycology and heavy metal pollution. *Biol Rev* 56: 99-151.
- Rai LC, Jansen TE, Rachlin JW (1989) A morphometric and X-ray energy dispersive approach to monitoring pH-altered cadmium toxicity in Anabaena flos-aquae. *Arch Environ Contam Toxicol* 19 (In Press).
- Rai LC, Raizada M (1986) Nickel induced stimulation of growth, heterocyst differentiation,  $^{14}\text{CO}_2$  fixation and nitrogenase activity of Nostoc muscorum. *New Phytol* 104: 111-114.
- Samson G, Morissette JC, Popovic R (1988) Copper quenching of the variable fluorescence in Dunaliella tertiolecta. New evidence for a copper inhibition effect on PS II photochemistry. *Photochem and Photobiol* 48: 329-332.
- Vierke G, Struckmeier P (1977) Binding in copper (II) to proteins of the photosynthetic membranes and its correlation with inhibition of electron transport in class II chloroplasts of Spinach. *Z Naturforsch* 32: 605-610.
- Vymazal J (1987) Toxicity and accumulation of cadmium with respect to algae and cyanobacteria: a review. *Toxicity Assess An Int Quat* 2: 387-415.
- Wong PTS (1987) Toxicity of cadmium to freshwater microorganisms, phytoplankton, and invertebrates In: Nriagu JO, Sprague JB (eds.) Cadmium in the Aquatic Environment, John Wiley & Sons Inc. pp 117-137.

Received February 22, 1990: accepted June 4, 1990.